ENANTIOSELECTIVE SYNTHESIS OF OPTICALLY PURE S-ISOSERINE

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<u>Abstract</u> : KF-promoted addition of nitromethane on (-)-8-phenylmenthyl glyoxylate monohydrate affords, after one purification, optically pure (-)S isoserine in about 50% yield.

Isoserine was found to inhibit several enzymes of serine metabolism in mammals and to interfer with panthotenic acid synthesis from β -alanine in yeast¹. More recently, it appeared to be a constituent of peptide antibiotics such as edeine² and tatumine³ and to

HO HO NH₂ (-)S isoserine

enhance antibiotic activity as in modified butirosin⁴ (where isoserine replaces 4-amino-2-hydroxy-butanoic acid). As such S-isoserine is thus an important biologically active β -amino acid.

Racemic isoserine has been synthesized in one step starting from α -chloro- β -hydroxypropionic acid⁵, or in two steps starting from methyl acrylate⁶, ethyl glycidate⁷ and glyoxylic acid hydrate⁸. S-isoserine has been prepared from S-asparagine⁹ or D-mannitol¹⁰, and R-isoserine from S-serine¹¹ or D-mannitol¹⁰.

We report here the first enantioselective synthesis of S-isoserine (which could also be used for the synthesis of R-isoserine). In this method, scheme 1, the inducer of chirality, (-)-8-phenyl menthol $\underline{1}^{12}$, is recovered and can thus be used again.

The key material is the (-)-8-phenylmenthol glyoxylate monohydrate $\underline{4}^{13}$ which is a stable colorless oil prepared in three steps from (-)-8-phenyl menthol $\underline{1}^{14}$. Using KF as a catalyst and anhydrous iPrOH as solvent¹⁵ nitromethane addition on $\underline{4}$ proceeds smoothly at 20°C¹⁶ leading to $\underline{5}$ in 90% yield. The use of KF to promote the condensation is also a key point of the synthesis as it allows the presence of an ester function on the substrate and, consequently, permits introduction of a chirality through a chiral ester¹⁷.

The percentage of asymmetric induction at C_{α} is determined by ¹H NMR (200MHz) on the <u>crude</u> products 5¹⁸ and 6¹⁹. It appears to be > 95% (fig. 1), as only one diastereomer is detected in both cases.



Scheme 1 : 1) HOCOCH₂Br/DCC/DMPA ; 2) AgNO₃/CH₃CN ; 3) AcONa/DMSO ; 4) CH₃NO₂/KF¹⁶ ; 5) protection ; 6) reduction ; 7) hydrolysis ; 8) epoxypropane/EtOH.

After protection (Et₃SiCl/DMF/Imidazole, overnight, rt) hydrogenation¹⁹ (Raney Ni, EtOH, H₂ 35 atm., 50°C, 24 h) and hydrolysis (HCl 6N, 80°C, 18 h) followed by treatment of the hydrochloride with epoxypropane in anhydrous ethanol, (-) S-isoserine²⁰, is isolated. Therefore, the diastereomer obtained in step 4 (where the new asymmetric carbon C_{α} is created) is 1R,2S,5R, $_{\alpha}$ S²¹.



Figure 1 : ¹H NMR (200MHz, CDCl₃) of crude product 5.

The observed strong shieldings of the α -H , -1.17 ppm, and of the two β -H, -0.88 and -0.64 ppm, (as compared with the same signals in the corresponding (-) menthyl ester²²), suggest a conformation, <u>A</u>, where the CH α -CH $_2\beta$ fragment is above the plane of the phenyl ring, i.e. cis to this phenyl group. The stability of this apparently sterically-hindered cis conformation, compared to the trans conformation <u>B</u>, could be due to the presence, in <u>A</u>, of an hyperconjugative stabilizing term of the type n₀(C-0-C).II,II*(phenyl) and/or II,II*(0-C0).II,II*(phenyl)²³.



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- 12) (-) 8-Phenylmenthol (1R,2S,5R) is prepared from (+) pulegone (E.J. Corey and H.E. Ensley, J. Am. Chem. Soc., <u>97</u>, 6908 (1975)). A 90(1R,2S,5R)/10(1S,2R,5R) mixture which is obtained, is flash-chromatographed (silicagel 60, ethyl ether/pentane 15/85).
- 13) <u>4</u> obtained in 70% overall yield from <u>1</u>, is more stable than the corresponding aldehyde and easier to handle.